High resolution interferometric FTIR spectroscopy of (HF)₂: analysis of a low frequency fundamental near 400 cm⁻¹

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The spectrum of $(HF)_2$ was recorded under equilibrium conditions in a long path thermostated cell at a resolution of 0.02 cm^{-1} between 350 and 550 cm⁻¹.

We present the first rotational analysis of a subband of the low frequency fundamentals of the HF dimer. The subband origins for the two tunnelling components of the $(v = 0, K = 0) \rightarrow (v = 1, K = 1)$ transition were found to be at 400.75₃₄ cm⁻¹ (B⁺ symmetry in the ground state) and at 399.78₆₅ cm⁻¹ (A⁺ symmetry in the ground state). The transition is tentatively assigned to the torsional vibration. The tunnelling frequency between the two equivalent isomers of (HF)₂ is 48.7₇ GHz for v = 1, K = 1, about a factor of 1.5 larger than in the ground state (v = 0, K = 1). The assignment of other subbands is also discussed.

1. Introduction

The dimer of hydrogen fluoride $(HF)_2$ is one of the simplest prototype molecules with hydrogen bonding [1]. Understanding its structure and dynamics will provide insight into detailed aspects of hydrogen bonding, which is important for a wide range of phenomena covering the structure of hydrogen bound liquids and their kinetics of evaporation and condensation, biological primary processes related to hydrogen bonding [1] and collisional energy transfer processes, say in the HF laser [2]. $(HF)_2$ has thus already been the subject of a variety of spectroscopic studies [3–17], including the vapour phase equilibria [18] and numerous theoretical studies of the potential energy surface [19–24].

From the pioneering studies in microwave spectroscopy [3–6] one knows that $(HF)_2$ is a planar, slightly asymmetric top with a nearly linear hydrogen bond and an average $F-H\cdots F$ distance of about 278 pm. The non H-bonded H atom is bent 63° off the $F\cdots F$ axis and executes a large amplitude exchange with the H-bonded H-atom leading to a tunnelling splitting of 0.6587 cm⁻¹ in the vibrational ground state (K = 0, for K = 1 : 1.064 cm⁻¹). The (HF)₂ molecule is thus also an interesting example for the application of non-rigid molecule group theory [25–29].

The high frequency HF stretching vibrations have been studied by Pine *et al.* [8, 9] at high resolution allowing both the rotational analysis and the determination of the predissociation linewidths, which are significantly mode dependent [8, 9] thus indicating mode selective predissociation behaviour [30]. We have been able to demonstrate that relatively narrow line structure is also visible in the HF stretching overtone spectrum of $(HF)_2$ [10, 11].

A careful search of the literature indicated a complete lack of high resolution analyses of the low frequency fundamentals of $(HF)_2$. These low frequency funda-

mentals are, however, of crucial importance for understanding both the potential surface for hydrogen bonding at large amplitudes of motion and for more detailed investigations of the predissociation dynamics. We thus started, some time ago [10, 11] a systematic investigation of the far infrared spectrum of $(HF)_2$. The present paper provides the first detailed account of the rovibrational analysis of a band near 400 cm⁻¹, which is tentatively assigned to the torsional fundamental transition.

2. Experimental

The spectra have been recorded between 350 and 550 cm⁻¹ on our BOMEM DA 002 interferometric Fourier transform spectrometer system allowing for a maximum resolving power of about 10^6 (apodized bandwidth $0.004 \,\mathrm{cm}^{-1}$). The spectra analysed below have been taken with an instrumental bandwidth of $0.02 \,\mathrm{cm}^{-1}$, optimizing the signal to noise ratio also in relation to pressure broadening. We used a home built thermostated stainless steel cell with transfer optics giving an optical path length of 10 m. (HF)₂ was created under equilibrium conditions between 250 K and 300 K at total pressures between 2 and 10 k Pa. The temperature was controlled to about ± 1 K. The signal strengths could be correlated with the approximately known (HF), partial pressures from the equilibrium constant. Unambiguous identification of the dimer spectrum is also available by means of combination differences (see below). HF was obtained from Matheson. There were no appreciable impurities other than small amounts of air and water. Some of our spectra showed the SiF₄ absorption near 389 cm^{-1} . This is probably due to the reaction of HF with Si-impurities in the polyethylene windows of the cell. The frequencies were calibrated against known frequencies for the HF monomer and for water [40]. The corrections were always small.

3. Results and discussion

3.1. The I.R.-spectrum of gaseous hydrogen fluoride between 350 and 550 cm⁻¹

Under equilibrium conditions the far infrared spectrum of hydrogen fluoride shows the prominent lines of the pure rotational spectrum of HF, extending to high frequencies, and in addition extended bands with very complex fine structure (looking like 'noise') and some broad background, which is difficult to distinguish from artifacts. Figure 1 shows a low resolution overview of the region between 395 and 505 cm^{-1} . The strongest feature assigned to (HF), appears at the low frequency end of the spectrum near 400 cm^{-1} , with structures extending to both low and high wavenumbers. As shown by the detailed assignment in figure 2, which is discussed below, this feature can be clearly assigned to $(HF)_2$ with two strong Q-branches and subband origins at 400.75₃₄ cm⁻¹ and 399.78₆₅ cm⁻¹. Further, weaker bands can be identified by band heads near $432 \cdot 2 \text{ cm}^{-1}$, $451 \cdot 3 \text{ cm}^{-1}$ and $471 \cdot 9 \text{ cm}^{-1}$. Even without detailed rotational assignments these can be identified as (HF)₂ absorptions, because of their pressure and temperature dependence. In contrast to this the obvious band with a maximum near 500.14 cm⁻¹ shows different pressure and temperature dependences as well as some other weak and unstructured absorptions, which are perhaps due to higher (HF), polymers.



Figure 1. Low resolution (0.3 cm^{-1}) survey spectrum of gaseous hydrogen fluoride, $p = 4 \text{ kPa}, l = 10 \text{ m}, T = 250 \text{ K}. \text{ HF}_{\text{mono}}$ indicates monomer rotational lines.

To low wavenumbers there is another intense band with Q-branch centres at $383 \cdot 34_4$ and $382 \cdot 02_8$ cm⁻¹. The R-branch regions of these bands overlap with P and Q-branch transitions of the band at 400 cm⁻¹ and are thus very congested and difficult to analyse. Before discussing the detailed rovibrational assignment, we shall summarize the current knowledge about the expected vibrational transitions at low frequencies of (HF)₂.

3.2 Preliminary discussion of the vibrational assignment

For the $(HF)_2$ molecule one has six vibrational fundamentals, five of A' species in the C_s point group and one of A" species in C_s (out-of-plane bending, hereafter called torsion). Two of the high frequency fundamentals are essentially HF stretching vibrations, which have been well studied [8, 9]. None of the low frequency vibrations has so far been analysed in detail. Table 1 summarizes theoretical predictions and table 2 some absorptions observed in low temperature matrix spectros-



Figure 2. The $K = 1 \leftarrow 0$ torsional subband of the (HF)₂ molecule (resolution 0.02 cm^{-1} FWHM, apodized) p = 2 kPa, T = 250 K, l = 10 m.

	ṽ/cm⁻	1				
$v_1(A')$	$v_2(A')$	$v_3(A')$	$v_4(A')$	$v_5(A')$	v ₆ (A")	of calculation
4279	4242	551	221	154	463	[19], HF/6–31
3905	3852	607	243	174	49 7	[19], MP2/6-31
4127	4054	582	231	163	516	[19], MP2/6-311
4070	4021	519	189	165	475	[20], DZ/SCF
3773	3743	523	196	166	459	[20], DZ/CI
4153	4095	607	218	156	486	[20], DZ + P/CI
4081	4038	588	226	171	519	[21], LCAO/SCF
3873	3753	697	333	212	623	[14], (H3)
3976¶	3920¶	420§	127§	157¶		[41]
	<u> </u>	520 [°]	178 [°]	337	same order	[23]
_	_	304	148	160	as v_3	[23]¶
3930-903†	3868-313†		_		399.78 5 \$	[9]†, this work‡
3930-459†	3867-421†	-	—	—	400·75 ₃₄ ‡	[9]†, this work‡

Table 1. Predicted§ and observed fundamental frequencies of $(HF)_2$.

[†] Ref. [9], upper line is for the symmetric tunnelling component in the ground state, the lower line for the antisymmetric component. Both are for the K' = K'' = 0 state.

[‡] This work, with the preliminary assignment to torsion, upper line for symmetric tunnelling component in the ground state, lower line for antisymmetric one. Both are assigned to the K' = 1, K'' = 0 state.

§ The predicted frequencies are mostly uncorrected for anharmonicity.

Uncorrected for anharmonicity, from a semiempirical potential surface model.

¶ Corrected for anharmonicity.

copy. The four low frequency fundamentals can be grouped in a high frequency pair, predicted to fall between 300 cm^{-1} to 700 cm^{-1} (at most) and a low frequency pair, about 150 to 300 cm^{-1} . The A'' torsion is predicted to belong to the high frequency pair. The matrix spectra show absorptions at a number of frequencies in this range. The band at 400 cm^{-1} could thus be due either to the torsional v_6 or to the high frequency in plane bending fundamental v_3 . There are also, in principle, three possible overtone and combination transitions from the low frequency fundamentals that may fall in this frequency range. Two of these would involve excitation of the large amplitude bending vibration v_5 with much increased tunnelling splitting, which we exclude below. The third would be the overtone $2v_4$ of the dimer stretching vibration. This coordinate corresponds to dissociation along the F-H \cdots F axis, with a dissociation wavenumber of about 1600 cm⁻¹.

Table 2. Absorptions assigned to the (HF)₂ molecule in low resolution vapour phase spectroscopy and low temperature matrix spectroscopy.

	Waven	umber/cn	n ⁻¹	Remarks [ref.]
561	381 385 446; 400 400	263	189	Vapour [13] Vapour [12] Ar matrix [14] Ar matrix [16] Ar matrix [17]
523	441 410			CO matrix [14] Ne matrix [15]

Further evidence for the assignment arises from the rotational polarization of the transition. The in plane vibrations are A/B hybrids, whereas the torsion is of type C [32]. With an asymmetry parameter $\kappa \simeq -0.998$, we can use essentially symmetric top selection rules. The hydrogen bond stretching vibration and its overtones are polarized along the top axis, leading to parallel bands with weak Qbranches because of the very large A rotational constant and the low $K_a \equiv K$ quantum numbers contributing to the spectrum. All other low frequency fundamentals give rise to perpendicular bands with a wide K structure and strong Q-branches in the K subbands. The observed structure roughly matches a perpendicular like transition and would thus have to be assigned either to the torsional (A'', v_6) or to the high frequency in plane bending vibration (A', v_3) . The temperature dependent measurements do not indicate the possibility of a vibrational hot band transition for the dominant part of the spectrum.

3.3. Detailed rotational analysis

The (HF)₂ molecule is highly non rigid with large amplitude motions both by vibration and by rotation. This has consequences, which have been discussed before [3-9, 29] and will be outlined here briefly, in order to make the analysis and notation comprehensible. (HF)₂ exists in two equivalent planar conformations which may easily interconvert as shown in figure 3. The tunnelling splittings related to this interconversion are easily observed, of the order of 1 cm^{-1} , depending strongly on the rotational and vibrational state [3-9]. The molecular symmetry group [25] is of order 4 and its character table is given in table 3. (HF)₂ is planar and the group is thus the direct product of the space inversion group $S^* = \{E, E^*\}$ and the permutation group involving the interchange of the two HF units a and b, $\{E, (ab)\} \equiv \{E, (13)(24)\}$. We use the systematic notation of [26] indicating parity (\pm) explicitly in the symbols for the species. The relationship to the C_{2h} point group notation [29] related to the most probable trans-tunnelling path originally suggested by Mills is given in parentheses. Each A^+ (A') or A^- (A") level in C_s symmetry is split by tunnelling in a $(A, B)^+$ and $(A, B)^-$ pair. The optical electric dipole selection rules are

- (i) parity change $(+ \leftrightarrow -)$; and
- (ii) conservation of nuclear spin symmetry (i.e. $A \nleftrightarrow B$).

Figure 4 is a diagram which shows (for the band observed here) the tunnelling splittings, symmetry species, optical transitions and nuclear spin statistical weights, which are easily obtained from elementary considerations in this simple symmetry





Species	E	E	• ((ab) (13)(24)	(<i>ab</i>)* (13)(24)*	$(M_{s_4}) \uparrow S^*_{2,2}$			
$ \begin{array}{c} A^+ (A_g) \\ A^- (A_u) \\ B^+ (B_u) \\ B^- (B_g) \end{array} $	1 1 1 1	_	1 1 1 1	$ \begin{array}{c} 1 \\ 1 \\ -1 \\ -1 \\ -1 \end{array} $	$ \begin{array}{c} 1 \\ -1 \\ -1 \\ 1 \end{array} $	$A_1^+ + A_2^+ A_1^- + A_2^- B_1^+ + B_2^- B_1^- + B_2^-$			
	(b) Character table for $S_{2,2}$.								
Species		Ε	(13)	(24)	(13)(34)	$(S_{2,2}) \downarrow M_{S2}$			
$A \times A \equiv A$ $B \times B \equiv A$ $B \times A \equiv E$ $A \times B \equiv E$	1 ₁ 1 ₂ 3 ₁ 3 ₂	1 1 1 1	-1 -1 1		1 -1 -1	A A B B			

Table 3. (a) Character table for the molecular symmetry group M_{54} of $(HF)_2$.

group. The symmetric (A) nuclear spin functions combining with the A motional functions may have total nuclear spin I = 2, 1 or 0 (twice for the latter, total weight g = 10) and the antisymmetric (B) nuclear spin functions, combining with B levels, may have I = 1 (twice, total weight g = 6). Table 3 shows for completeness also the induced representations $\Gamma(M_{S4}) \uparrow S_{2,2}^*$ for the full permutation inversion group.



Figure 4. Scheme for the observed rovibrational transitions assigned to the torsional fundamental of $(HF)_2 \Gamma^{\nu b} \Gamma^r = \Gamma^{\nu r b}$ is the vibrational, rotational species *including* the bending motion with the tunneling splitting. Γ^r is the rotational species.

Only the character table for $S_{2,2}$ is defined and parity is simply to be added. The splittings in these doublets $A_1 + A_2$ and $B_1 + B_2$ with either + or - parity are related to the tunnelling, if proton transfer across the hydrogen bond and consequently also of the second proton were to become important. No such splittings have been observed to date and they are estimated to be quite small. One might speculate that the splitting may become appreciable at very high overtones for the HF stretching vibration, but we have found no evidence for this [33]. These considerations define thus the level structure arising from tunnelling.

The second pecularity due to large amplitude motion arises from the rotation around the symmetric top *a*-axis (about $F \cdots H$ -F line), which has an extremely large rotational 'constant' A and centrifugal perturbation as is obvious already from chemical intuition. As discussed in detail by Pine *et al.* [8, 9] it is best to treat every $K = K_a$ level with an independent set of constants and the approximate term formula

$$\frac{E_{vK}}{hc} = \tilde{v}_{vK} + \bar{B}_{vK}J(J+1) - D_{vK}[J(J+1) - K^2]^2 + \frac{\Delta E_{vK}}{hc}, \qquad (1 a)$$

$$\frac{\Delta E_{vK}}{hc} = \pm \delta_{K1} \left[\frac{1}{4} (B_{vK} - C_{vK}) J (J+1) + \frac{1}{2} l_{vK} J^2 (J+1)^2 \right]. \tag{1b}$$

 \tilde{v}_{vK} is the subband origin of the K transition and $\bar{B}_{vK} = (B_{vK} + C_{vK})/2$ is the average rotational constant. E_{vK} is a near symmetric top approximation with an asymmetry splitting ΔE_{vK} , which is appreciable only for K = 1 (Kronecker delta δ_{K1} in equation (1 b)). The + sign in equation (1 b) corresponds to $K_a + K_c = J$ and the - sign to $K_a + K_c = J + 1$. D_{vK} and l_{vK} are the appropriate centrifugal constants. Higher order corrections could be neglected in our fits. These fits were extended to values of J'' = 31, converging well, using the Marquardt algorithm [34].

Table 4 summarizes the lines which could be assigned to the first tunnelling component with a subband origin at 400.75₃₄ cm⁻¹. The values in parentheses are of lines broadened by overlapping with other lines or are otherwise uncertain and were not used in the fit. The J-assignment is quite well established. When fitting the Q- and R-branches separately we obtained with the J-assignment of table 4 the same subband origin for both fits within the standard deviation (about a tenth of the resolution). A change of the J-assignment by ± 1 for the Q-branch leads to a discrepancy by about twice the resolution, i.e. a factor of 20 poorer, which is significant. The lowest J", which could be assigned is 2, lower values leading to weak and overlapping bands. Lines with odd J" are more intense. This identifies the assignment of the vibronic symmetry of the lower state of the transition as $\Gamma^{vb''} = B^+$ (we write Γ^{vb} in order to specify vibronic symmetry species including the bending motion giving the tunnelling splitting using M_{S4} , see also [29]). From combination differences we can establish that the lower vibrational state of the transition is the vibrational ground state (except for the tunnelling splitting)

$$\Delta_2 F(J_{K_aK_c}') = F''(J+1) - F''(J-1) = R(J-1) - P(J+1).$$
⁽²⁾

These values are summarized in table 5 and are compared with accurate data from microwave [3-6] and near-IR laser spectroscopy [8, 9]. The root mean square deviation for the best assignment is about 20 per cent of the resolution. An accidental agreement with rotational levels from an excited vibrational state (except for tunnelling) can be safely excluded at this level. The rotational K assignment can also

Table 4. Assigned transitions of the band with subband origin at 400.75_{34} cm⁻¹ (values in parentheses were not included in the fit). The ground level constants were fixed in the fit. The root mean square deviation is 0.004 cm⁻¹ for 51 lines with 5 parameters. (The wavenumbers are all calibrated and corrected.)

		Wavenumber/cm ⁻¹			
<i>J</i> ′	J ″	ν̃ _{exp}	$(\tilde{v}_{exp} - \tilde{v}_{calc})$		
5	6	(397.965)	-0.0068		
6	7	(397-461)	-0.0047		
9	10	(395-871)	-0.0026		
10	11	395.3182	0.0000		
11	12	394.7519	0.0016		
12	13	394-1665	-0.0033		
13	14	(393-550)	-0.0266		
14	15	(392.956)	-0.0145		
16	17	391.7107	-0.0084		
17	18	391.0769	0.0034		
18	19	390-4143	0.0000		
19	20	389.7494	0.0080		
20	21	389.0511	-0.0034		
22	23	387.6438	0.0059		
23	24	386-9050	-0.0026		
24	25	(386-175)	0.0126		
26	27	(384-617)	-0.0088		
27	28	383.8380	0.0042		
28	29	383.0208	-0.0048		
29	30	(382-184)	-0.0168		
2	2	(400.721)	0.0060		
3	3	400-6727	-0.0038		
4	4	400-6283	0.0031		
5	5	400-5639	0.0028		
6	6	400-4780	-0.0059		
7	7	400-3961	0.0023		
8	8	400·2914	0.0009		
9	9	400·1743	0.0001		
10	10	400.0443	-0.0005		
11	11	399.9013	-0.0005		
12	12	399·7415	-0.0036		
13	13	399.5794	0.0043		
14	14	399-3851	-0.0063		
15	15	399.1927	-0.0011		
16	16	(398-978)	-0.0042		
17	17	398.7596	0.0031		
18	18	398-5176	0.0012		
19	19	398-2561	-0.0057		
20	20	397·9959	0.0034		
21	21	397.7085	0.0002		
22	22	397.4073	-0.0012		
23	23	397.0974	0.0030		
24	24	396.7696	0.0054		
25	25	(396·414)	-0.0045		
26	26	396-0513	-0.0048		
27	27	(395.671)	-0.0068		
28	28	(395-274)	-0.0088		
29	29	(394.856)	-0·0150		

		Wavenumber/cm ⁻¹				
_ J '	J″	ν̃ _{exp}	$(\tilde{v}_{exp} - \tilde{v}_{calc})$			
4	3	402.3656	0.0030			
6	5	403.0969	0.0049			
7	6	403.4454	0.0079			
8	7	403.7654	0.0046			
9	8	404.0943	0.0020			
10	9	(404·389)	0.0063			
11	10	404.6830	0.0048			
12	11	404.9628	-0.0037			
13	12	(405.244)	0.0130			
14	13	`405 ∙4776	0.0043			
15	14	405.7171	-0.0009			
16	15	(405.952)	0.0129			
17	16	406-1437	0.0023			
18	17	(406-327)	-0.0100			
19	18	406.5143	0.0013			
20	19	406.6704	-0.0024			
21	20	406.8122	0.0042			
22	21	406-9454	0.0019			
23	22	407.0613	0.0076			
24	23	407.1488	0.0023			
25	24	(407-235)	0.0135			
26	25	(407.267)	-0.0119			
27	26	(407-314)	-0.0035			

Table 4 (continued).

be established on the basis of our data. K > 2 can be excluded at once because of the assigned J'' = 2. K doubling is not observed, although it might possibly occur for a $K = 1 \leftarrow 2$ transition. The observation of intensity alternation excludes $K_a \ge 2$ because $K_c =$ even or odd are not split appreciably for these K_a . Hence, only K = 0or 1 need be considered. Table 5 gives root mean square deviation for various K assignments. It is evident that only K'' = 0 is consistent with the data. The assignment of the tunnelling components of the vibrational ground level is somewhat less obvious from the frequency data alone.

Table 6 gives the assigned transitions for the second tunnelling component with a subband origin at 399.78₆₅ cm⁻¹. Here, especially the Q-branch transitions from the lower J" overlap with transitions of the 400.75₅₄ band. Using nuclear spin statistics and the other procedures discussed above, we conclude that the band is a transition from K'' = 0 in the ground state to K' = 1 in the excited state. Q-branch transitions go to the lower, P and R-branch transitions to the upper level of the asymmetry split $K_a = 1$ vibrationally excited state. This is true for both subbands and is illustrated in figure 4. Nuclear spin statistics establish the vibronic species of the lower state to be $\Gamma^{vb''} = A^+$.

Table 7 summarizes the spectroscopic constants obtained from the fits for the two subbands. Figure 5 shows a detail of the comparison of the experimental and simulated spectrum. Noting that some congestion from weak hot band transitions is expected, the agreement is very good.

				$\Delta_2 F''/c$	cm ^{- 1}				
					Mic	rowave d	ata§		
J	$\Delta_2 F''(J)^{\dagger}$	$\Delta_2 F''(J)$ ‡	J _{KaKc}	$\Delta_2 F''(J) \\ A^+$	$\begin{array}{c}\Delta_2 \ F''(J)\\B^+\end{array}$	J _{KaKc}	$\frac{\Delta_2 F''(J)}{A^+}$	J _{KaKc}	$\frac{\Delta_2 F''(J)}{A^+}$
10	9.0793		10,10	9.0825	9.0771	10, 10	9.0995	10, .	9.1654
11		9.9311	11_{011}	9.9433	9.9374	11, 11	9.9621	11,10	10.0340
12		10.7963	12_{012}	10.8030	10.7965	12_{112}	10.8235	12, 11	10.9015
15	13-3752		15015	13-3741	13.3662	15, 15	13.4000	15,14	13.4960
17	15.0735	15-0668	17017	15.0804	15.0717	17, 17	15-1103	17, 16	15·2179
19		16.7649	19019	16·7797	16.7702	19, 19	16.8136	19118	16-9326
20	17.6276	17.6193	20_{020}	17.6263	17.6165	$20_{1,20}$	17.6624	20, 19	17.7871
21	18·4613		21_{021}	18-4709	18.4607	21_{121}	18.5092	21_{120}	18.6394
22	19·3173	19·3016	22_{022}	19-3132	19.3027	22_{122}	19.3543	221 21	19.4894
23	20.1503	20.1563	23_{023}	20.1532	20.1424	23_{123}	20·1962	23_{122}	20.3371
25	21.8324		250 25	21.8257	21.8144	251 25	21.8737	25_{124}	22·0251
27		23.4760	27 ₀₂₇	23.4875	23.4758	271 27	23.5408	271 26	23.7023
rms :	$\tilde{v} = 1$	399,7865 cm	-1	0.0053	0.010		0.0372		0.1588
	$\tilde{v} = 4$	400,7534 cm	- 1	0.0107	0.0061		0.0453		0.1658

Table 5. Ground state combination differences $\Delta_2 F''(J)$ (see also text).

† For the band with subband centre at 399.7865 cm^{-1} .

 \ddagger For the band with subband centre at 400.7534 cm⁻¹.

§ From [3, 4, 5, 8, 9].

|| The root mean square deviation is given for possible assignments of the subbands to different transitions as indicated. rms² = $N^{-1} \sum_{i=1}^{N} (\Delta_2 F_i'' - \Delta_2 F_{i(microwave)}')^2$. N = number of assigned differences.

Table 6. Assigned transitions of the band with subband origin at 399.78₆₅ cm⁻¹. The root mean square deviation for 43 lines and 5 parameters is 0.0031 cm⁻¹ (see also table 4).

		Wavenumber/cm ⁻¹				
J′	J″	ν̃ _{exp}	$(\tilde{v}_{exp} - \tilde{v}_{calc})$			
8	9	(395.477)	0.0162			
9	10	394-924	0.0030			
10	11	394 ·3661	-0.0032			
11	12	393-8100	0.0044			
12	13	393-2269	-0.0029			
13	14	(392.637)	-0.0048			
15	16	391.4286	-0.0005			
16	17	(390-806)	0.0025			
17	18	390-1708	0.0052			
20	21	388.1758	0.0018			
21	22	387-4911	0.0074			
22	23	386.7798	0.0000			
23	24	386.0637	0.0016			
24	25	(385-342)	0.0116			
25	26	384-5833	-0.0011			
26	27	(383-840)	0.0161			
27	28	(383-045)	-0.0037			

J' J'' \tilde{v}_{exp} $(\tilde{v}_{exp} - \tilde{v}_{calc})$ 2 2 (399.743) -0.0055 4 4 (399.662) 0.0023 5 5 (399.584) -0.0122 6 6 (399.531) 0.0111 7 7 (399.432) 0.00057 10 10 399.0863 -0.0022 9 9 (399.219) 0.0057 10 10 399.0863 0.0014 11 11 (398.956) 0.0126 12 12 398.7860 -0.0025 13 13 398.6188 -0.0013 14 14 398.4429 0.0047 15 15 (398.256) 0.0137 16 16 398.0333 0.0005 17 17 397.8053 -0.0042 20 20 397.0550 0.0029 21 21 396.7696 -0.00015			Wavenur	mber/cm ⁻¹
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	J′		ν̃ _{exp}	$(\tilde{v}_{exp} - \tilde{v}_{calc})$
44 $(399\cdot662)$ $0\cdot0023$ 55 $(399\cdot584)$ $-0\cdot0122$ 66 $(399\cdot531)$ $0\cdot0111$ 77 $(399\cdot432)$ $0\cdot0016$ 88 $399\cdot3263$ $-0\cdot0022$ 99 $(399\cdot219)$ $0\cdot0057$ 1010 $399\cdot0863$ $0\cdot0014$ 1111 $(398\cdot956)$ $0\cdot0126$ 1212 $398\cdot7860$ $-0\cdot0025$ 1313 $398\cdot6188$ $-0\cdot0013$ 1414 $398\cdot4429$ $0\cdot0047$ 1515 $(398\cdot256)$ $0\cdot0137$ 1616 $398\cdot0333$ $0\cdot0005$ 1717 $397\cdot8053$ $-0\cdot0039$ 1818 $397\cdot5671$ $-0\cdot0042$ 2020 $397\cdot0550$ $0\cdot0029$ 2121 $396\cdot7696$ $-0\cdot0007$ 2222 $396\cdot1630$ $0\cdot0015$ 2424 $395\cdot8385$ $0\cdot0045$ 2525 $(395\cdot473)$ $-0\cdot0175$ 2626 $395\cdot1329$ $0\cdot0014$ 2727 $394\cdot7519$ $-0\cdot0041$ 2828 $(394\cdot366)$ $0\cdot021$ 2929 $(393\cdot938)$ $-0\cdot0170$ 3030 $(393\cdot529)$ $0\cdot0000$ 3131 $(393\cdot071)$ $-0\cdot0145$ 32 $(401\cdot014)$ $-0\cdot0011$ 43 $401\cdot7767$ $0\cdot0031$ 65 $(402\cdot121)$ $-0\cdot0135$ 76 $402\cdot4864$ $0\cdot0035$ 98 </td <td>2</td> <td>2</td> <td>(399.743)</td> <td>-0.0055</td>	2	2	(399.743)	-0.0055
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	4	4	(399.662)	0.0023
66 $(399 \cdot 531)$ $0 \cdot 0111$ 77 $(399 \cdot 432)$ $0 \cdot 0016$ 88 $399 \cdot 3263$ $-0 \cdot 0022$ 99 $(399 \cdot 219)$ $0 \cdot 0057$ 1010 $399 \cdot 0863$ $0 \cdot 0014$ 1111 $(398 \cdot 956)$ $0 \cdot 0126$ 1212 $398 \cdot 7860$ $-0 \cdot 0025$ 1313 $398 \cdot 6188$ $-0 \cdot 0013$ 1414 $398 \cdot 4429$ $0 \cdot 0047$ 1515 $(398 \cdot 256)$ $0 \cdot 0137$ 1616 $398 \cdot 0333$ $-0 \cdot 0005$ 1717 $397 \cdot 8053$ $-0 \cdot 0039$ 1818 $397 \cdot 5671$ $-0 \cdot 00422$ 2020 $397 \cdot 0550$ $0 \cdot 0029$ 2121 $396 \cdot 7696$ $-0 \cdot 0007$ 2222 $396 \cdot 4732$ $-0 \cdot 00042$ 2323 $396 \cdot 1630$ $0 \cdot 0015$ 2424 $395 \cdot 8385$ $0 \cdot 0045$ 2525 $(395 \cdot 473)$ $-0 \cdot 0175$ 2626 $395 \cdot 1329$ $0 \cdot 00014$ 2727 $394 \cdot 7519$ $-0 \cdot 00041$ 2828 $(394 \cdot 366)$ $0 \cdot 0021$ 2929 $(393 \cdot 928)$ $-0 \cdot 0170$ 3030 $(393 \cdot 529)$ $0 \cdot 0000$ 3131 $(393 \cdot 071)$ $-0 \cdot 0145$ 32 $(401 \cdot 014)$ $-0 \cdot 0011$ 43 $401 \cdot 4001$ $-0 \cdot 0031$ 65 $(402 \cdot 121)$ $-0 \cdot 0135$ 76 402	5	5	(399.584)	-0.0122
77 $(399\cdot432)$ $0\cdot0016$ 88 $399\cdot3263$ $-0\cdot0022$ 99 $(399\cdot219)$ $0\cdot0057$ 1010 $399\cdot0863$ $0\cdot0014$ 1111 $(398\cdot956)$ $0\cdot0126$ 1212 $398\cdot7860$ $-0\cdot0025$ 1313 $398\cdot6188$ $-0\cdot0013$ 1414 $398\cdot4429$ $0\cdot0047$ 1515 $(398\cdot256)$ $0\cdot0137$ 1616 $398\cdot0333$ $0\cdot0005$ 1717 $397\cdot8053$ $-0\cdot0039$ 1818 $397\cdot5671$ $-0\cdot0042$ 2020 $397\cdot0550$ $0\cdot0029$ 2121 $396\cdot7696$ $-0\cdot0007$ 2222 $396\cdot4732$ $-0\cdot00042$ 2323 $396\cdot1630$ $0\cdot0015$ 2424 $395\cdot8385$ $0\cdot0045$ 2525 $(395\cdot473)$ $-0\cdot0175$ 2626 $395\cdot1329$ $0\cdot0014$ 2727 $394\cdot7519$ $-0\cdot0041$ 2828 $(394\cdot366)$ $0\cdot0021$ 2929 $(393\cdot938)$ $-0\cdot0170$ 3030 $(393\cdot529)$ $0\cdot0000$ 3131 $(393\cdot071)$ $-0\cdot0114$ 43 $401\cdot014$ $-0\cdot0011$ 43 $401\cdot014$ $-0\cdot0031$ 65 $(402\cdot121)$ $-0\cdot0135$ 76 $402\cdot4864$ $0\cdot0035$ 98 $403\cdot1439$ $0\cdot0022$ 109 $403\cdot4454$ $-0\cdot0065$ 111	6	6	(399·531)	0.0111
8 8 $399 \cdot 3263$ -0.0022 9 9 $(399 \cdot 219)$ 0.0057 10 10 $399 \cdot 0863$ 0.0014 11 11 $(398 \cdot 956)$ 0.0126 12 12 $398 \cdot 7860$ -0.0025 13 13 $398 \cdot 6188$ -0.0013 14 14 $398 \cdot 4429$ 0.0047 15 15 $(398 \cdot 256)$ 0.0137 16 16 $398 \cdot 0333$ 0.0005 17 17 $397 \cdot 8053$ -0.0039 18 18 $397 \cdot 5671$ -0.00422 20 20 $397 \cdot 0550$ 0.0029 21 21 $396 \cdot 7696$ -0.0007 22 22 $396 \cdot 4732$ -0.000442 23 23 $396 \cdot 1329$ 0.00145 24 $395 \cdot 8385$ 0.0045 25 25 $(395 \cdot 473)$ -0.0175 26 26 $395 \cdot 1329$ 0.0014 27 27 $394 \cdot 7519$ -0.00041	7	7	(399-432)	0.0016
99 $(399 \cdot 219)$ $0 \cdot 0057$ 1010 $399 \cdot 0863$ $0 \cdot 0014$ 1111 $(398 \cdot 956)$ $0 \cdot 0126$ 1212 $398 \cdot 7860$ $-0 \cdot 0025$ 1313 $398 \cdot 6188$ $-0 \cdot 0013$ 1414 $398 \cdot 4429$ $0 \cdot 0047$ 1515 $(398 \cdot 256)$ $0 \cdot 0137$ 1616 $398 \cdot 0333$ $-0 \cdot 0005$ 1717 $397 \cdot 8053$ $-0 \cdot 0039$ 1818 $397 \cdot 5671$ $-0 \cdot 00422$ 2020 $397 \cdot 0550$ $0 \cdot 0029$ 2121 $396 \cdot 7696$ $-0 \cdot 0007$ 2222 $396 \cdot 4732$ $-0 \cdot 0004423$ 2323 $396 \cdot 1630$ $0 \cdot 0015$ 2424 $395 \cdot 8385$ $0 \cdot 00455$ 2525 $(395 \cdot 473)$ $-0 \cdot 0175$ 2626 $395 \cdot 1329$ $0 \cdot 00014$ 2729 $(393 \cdot 938)$ $-0 \cdot 0170$ 3030 $(393 \cdot 529)$ $0 \cdot 0000$ 3131 $(393 \cdot 071)$ $-0 \cdot 0111$ 43 $401 \cdot 014$ $-0 \cdot 0031$ 65 $(402 \cdot 121)$ $-0 \cdot 0135$ 76 $402 \cdot 4864$ $0 \cdot 0035$ 98 $403 \cdot 1439$ $0 \cdot 0022$ 10 $\cdot 9$ $403 \cdot 4454$ $-0 \cdot 0065$ 1110 $(403 \cdot 765)$ $0 \cdot 0163$ 1312 $404 \cdot 3008$ $-0 \cdot 0027$ 1413 $404 \cdot 5608$ $0 \cdot 0022$ 10 $\cdot 9$ <td>8</td> <td>8</td> <td>399.3263</td> <td>-0.0022</td>	8	8	399.3263	-0.0022
1010399.0863 0.0014 1111(398.956) 0.0126 1212398.7860 -0.0025 1313398.6188 -0.0013 1414398.4429 0.0047 1515(398.256) 0.0137 1616398.0333 0.0005 1717397.8053 -0.0039 1818397.5671 -0.0042 2020397.0550 0.0029 2121396.7696 -0.0007 2222396.4732 -0.0004 2323396.1630 0.0015 2424395.8385 0.0045 2525(395.473) -0.0175 2626395.1329 0.0014 2727394.7519 -0.0041 2828(394.366) 0.0021 2929(393.938) -0.0170 3030(393.529) 0.0000 3131(393.071) -0.0145 32(401.014) -0.0011 43401.4001 -0.0031 65(402.121) -0.0135 76402.4864 0.0035 98403.1439 0.0022 109403.4454 -0.0065 1110(403.765) 0.0163 1312404.3008 -0.0027 1413404.5608 0.0005 1514404.8038 0.0005	9	9	(399-219)	0.0057
1111 $(398 \cdot 956)$ $0 \cdot 0126$ 1212 $398 \cdot 7860$ $-0 \cdot 0025$ 1313 $398 \cdot 6188$ $-0 \cdot 0013$ 1414 $398 \cdot 4429$ $0 \cdot 0047$ 1515 $(398 \cdot 256)$ $0 \cdot 0137$ 1616 $398 \cdot 0333$ $0 \cdot 0005$ 1717 $397 \cdot 8053$ $-0 \cdot 0039$ 1818 $397 \cdot 5671$ $-0 \cdot 0042$ 2020 $397 \cdot 0550$ $0 \cdot 0029$ 2121 $396 \cdot 7696$ $-0 \cdot 0007$ 2222 $396 \cdot 4732$ $-0 \cdot 00044$ 2323 $396 \cdot 1630$ $0 \cdot 0015$ 2424 $395 \cdot 8385$ $0 \cdot 0045$ 2525 $(395 \cdot 473)$ $-0 \cdot 0175$ 2626 $395 \cdot 1329$ $0 \cdot 0014$ 2727 $394 \cdot 7519$ $-0 \cdot 0014$ 28 $(394 \cdot 366)$ $0 \cdot 0021$ 2929 $(393 \cdot 529)$ $0 \cdot 0000$ 3131 $(393 \cdot 071)$ $-0 \cdot 0111$ 43 $401 \cdot 014$ $-0 \cdot 0011$ 43 $401 \cdot 014$ $-0 \cdot 0031$ 65 $(402 \cdot 121)$ $-0 \cdot 0135$ 76 $402 \cdot 4864$ $0 \cdot 0022$ 10 $\cdot 9$ $403 \cdot 4454$ $-0 \cdot 0065$ 1110 $(403 \cdot 765)$ $0 \cdot 0163$ 1312 $404 \cdot 3008$ $-0 \cdot 0027$ 1413 $404 \cdot 5608$ $0 \cdot 0027$ 1413 $405 \cdot 6313$ $-0 \cdot 0026$ 1918 $405 \cdot 6313$ <	10	10	399.0863	0.0014
1212398.7860 -0.0025 1313398.6188 -0.0013 1414398.4429 0.0047 1515(398.256) 0.0137 1616398.0333 0.0005 1717397.8053 -0.0039 1818397.5671 -0.0042 2020397.0550 0.0029 2121396.7696 -0.0007 2222396.4732 -0.0004 2323396.1630 0.0015 2424395.8385 0.0045 2525(395.473) -0.0175 2626395.1329 0.0014 2727394.7519 -0.0041 2828(394.366) 0.0021 2929(393.529) 0.0000 3131(393.071) -0.0145 32(401.014) -0.0011 43401.4001 -0.0004 54401.7767 0.0031 65(402.121) -0.0135 76402.4864 0.0035 98403.1439 0.0022 109403.4454 -0.0065 1110(403.765) 0.0163 1312404.3008 -0.0027 1413404.5608 0.0005 1514405.6313 -0.0026 1918405.6313 -0.0047 2221406.9771 0.0015 24<	11	11	(398-956)	0.0126
1313398.6188 -0.0013 1414398.4429 0.0047 1515(398.256) 0.0137 1616398.0333 0.0005 1717397.8053 -0.0039 1818397.5671 -0.0042 2020397.0550 0.0029 2121396.7696 -0.0007 2222396.4732 -0.0004 2323396.1630 0.0015 2424395.8385 0.0045 2525(395.473) -0.0175 2626395.1329 0.0014 2727394.7519 -0.0041 2828(394.366) 0.0021 2929(393.529) 0.0000 3131(393.071) -0.0145 32(401.014) -0.00111 43401.4001 -0.0004 54401.7767 0.0031 65(402.121) -0.0135 76402.4864 0.0035 98403.1439 0.0022 109403.4454 -0.0065 1110(403.765) 0.0163 1312404.3008 -0.0027 1413404.5608 0.0005 1514405.6313 -0.0038 1716405.2443 -0.0026 1918405.6313 -0.0047 2221406.9771 0.0011 21	12	12	398.7860	-0.0022
1414398·4429 0.0047 1515 $(398\cdot256)$ 0.0137 1616398·0333 0.0005 1717397·8053 -0.0039 1818397·5671 -0.0042 2020397·0550 0.0029 2121396·7696 -0.0007 2222396·4732 -0.0004 2323396·1630 0.0015 2424395·8385 0.0045 2525 $(395\cdot473)$ -0.0175 2626395·1329 0.0014 2727394·7519 -0.0041 2828 $(394\cdot366)$ 0.0021 2929 $(393\cdot529)$ 0.0000 3131 $(393\cdot071)$ -0.0145 32 $(401\cdot014)$ -0.00111 43 $401\cdot4001$ -0.0031 65 $(402\cdot121)$ -0.0135 76 $402\cdot4864$ 0.0035 98 $403\cdot1439$ 0.0022 109 $403\cdot4454$ -0.0065 1110 $(403\cdot765)$ 0.0163 1312 $404\cdot3008$ -0.0027 1413 $404\cdot5608$ 0.0005 1514 $404\cdot8038$ 0.0005 1615 $405\cdot2443$ -0.0026 1918 $405\cdot6313$ -0.0047 2221 $406\cdot971$ 0.0011 2120 $405\cdot9524$ -0.0047 2221 $406\cdot97$	13	13	398-6188	-0.0013
1515 $(398 \cdot 256)$ 0.0137 1616 $398 \cdot 0333$ 0.0005 1717 $397 \cdot 8053$ -0.0039 1818 $397 \cdot 5671$ -0.0042 2020 $397 \cdot 0550$ 0.0029 2121 $396 \cdot 7696$ -0.0007 2222 $396 \cdot 4732$ -0.0004 2323 $396 \cdot 1630$ 0.0015 2424 $395 \cdot 8385$ 0.0045 2525 $(395 \cdot 473)$ -0.0175 2626 $395 \cdot 1329$ 0.0014 2727 $394 \cdot 7519$ -0.0041 2828 $(394 \cdot 366)$ 0.0021 2929 $(393 \cdot 529)$ 0.0000 3131 $(393 \cdot 071)$ -0.0145 32 $(401 \cdot 014)$ -0.00111 43 $401 \cdot 4001$ -0.0004 54 $401 \cdot 7767$ 0.0031 65 $(402 \cdot 121)$ -0.0135 76 $402 \cdot 4864$ 0.0035 98 $403 \cdot 1439$ 0.0022 10 $\cdot 9$ $403 \cdot 4454$ -0.0065 1110 $(403 \cdot 765)$ 0.0163 1312 $404 \cdot 3008$ -0.0027 1413 $404 \cdot 5608$ 0.0005 1514 $404 \cdot 8038$ 0.0005 1615 $405 \cdot 2443$ -0.0026 1918 $405 \cdot 6313$ -0.0047 2221 $406 \cdot 9771$ 0.0011 2120<	14	14	398.4429	0.0047
1616 $398 \cdot 0333$ $0 \cdot 0005$ 1717 $397 \cdot 8053$ $-0 \cdot 0039$ 1818 $397 \cdot 5671$ $-0 \cdot 0042$ 2020 $397 \cdot 0550$ $0 \cdot 0029$ 2121 $396 \cdot 7696$ $-0 \cdot 0007$ 2222 $396 \cdot 4732$ $-0 \cdot 0004$ 2323 $396 \cdot 1630$ $0 \cdot 0015$ 2424 $395 \cdot 8385$ $0 \cdot 0045$ 2525 $(395 \cdot 473)$ $-0 \cdot 0175$ 2626 $395 \cdot 1329$ $0 \cdot 0014$ 2727 $394 \cdot 7519$ $-0 \cdot 0041$ 2828 $(394 \cdot 366)$ $0 \cdot 0021$ 2929 $(393 \cdot 529)$ $0 \cdot 0000$ 3131 $(393 \cdot 071)$ $-0 \cdot 0145$ 32 $(401 \cdot 014)$ $-0 \cdot 00111$ 43 $401 \cdot 4001$ $-0 \cdot 0004$ 54 $401 \cdot 7767$ $0 \cdot 0031$ 65 $(402 \cdot 121)$ $-0 \cdot 0135$ 76 $402 \cdot 4864$ $0 \cdot 0035$ 98 $403 \cdot 1439$ $0 \cdot 0022$ 10 $\cdot 9$ $403 \cdot 4454$ $-0 \cdot 0065$ 1110 $(403 \cdot 765)$ $0 \cdot 0163$ 1312 $404 \cdot 3008$ $-0 \cdot 0027$ 1413 $404 \cdot 5608$ $0 \cdot 0005$ 1514 $404 \cdot 8038$ $0 \cdot 0005$ 1615 $405 \cdot 2443$ $-0 \cdot 0026$ 1918 $405 \cdot 6313$ $-0 \cdot 0026$ 1918 $405 \cdot 6313$ $-0 \cdot 0047$ 2221 $406 $	15	15	(398-256)	0.0137
17 17 397.8053 -0.0039 18 18 397.5671 -0.0042 20 20 397.0550 0.0029 21 21 396.7696 -0.0007 22 22 396.4732 -0.0004 23 23 396.1630 0.0015 24 24 395.8385 0.0045 25 25 (395.473) -0.0175 26 26 395.1329 0.0014 27 27 394.7519 -0.0041 28 28 (394.366) 0.0021 29 29 (393.938) -0.0170 30 30 (393.529) 0.0000 31 31 (393.071) -0.0145 3 2 (401.014) -0.00111 4 3 401.4001 -0.0004 5 4 401.7767 0.0031 6 5 (402.121) -0.0135 7 6 402.4864 0.0035 9 8 403.1439 0.0022 10 9 403.4454 -0.0065 11 10 (403.765) 0.0163 13 12 404.3008 -0.0027 14 13 404.5608 0.0005 15 14 404.8038 0.0005 16 15 405.0285 -0.0038 17 16 405.2443 -0.0026 19 18 405.6313 -0.0047 22 21 40	16	16	398·0333	0.0005
1818 $397 \cdot 5671$ -0.0042 2020 $397 \cdot 0550$ 0.0029 2121 $396 \cdot 7696$ -0.0007 2222 $396 \cdot 4732$ -0.0004 2323 $396 \cdot 1630$ 0.0015 2424 $395 \cdot 8385$ 0.0045 2525 $(395 \cdot 473)$ -0.0175 2626 $395 \cdot 1329$ 0.0014 2727 $394 \cdot 7519$ -0.00411 2828 $(394 \cdot 366)$ 0.00211 2929 $(393 \cdot 529)$ 0.0000 3131 $(393 \cdot 071)$ -0.0145 32 $(401 \cdot 014)$ -0.00111 43 $401 \cdot 4001$ -0.0004 54 $401 \cdot 7767$ 0.0031 65 $(402 \cdot 121)$ -0.0135 76 $402 \cdot 4864$ 0.0035 98 $403 \cdot 1439$ 0.0022 109 $403 \cdot 4454$ -0.0065 1110 $(403 \cdot 765)$ 0.0163 1312 $404 \cdot 3008$ -0.0027 1413 $404 \cdot 5608$ 0.0005 1514 $404 \cdot 8038$ 0.0005 1615 $405 \cdot 2443$ -0.0026 1918 $405 \cdot 6313$ -0.0047 2221 $406 \cdot 0971$ 0.0011 2120 $405 \cdot 9524$ -0.0047 2221 $406 \cdot 2140$ -0.0051 2423 $406 \cdot 3273$ 0.0015	17	17	397.8053	-0·0039
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	18	18	397.5671	-0.0042
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	20	20	397 ·0550	0.0029
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	21	21	396.7696	-0.0001
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	22	22	396.4732	- 0.0004
24 24 395.8385 0.0045 25 25 (395.473) -0.0175 26 26 395.1329 0.0014 27 27 394.7519 -0.0041 28 28 (394.366) 0.0021 29 29 (393.938) -0.0170 30 30 (393.529) 0.0000 31 31 (393.071) -0.0145 3 2 (401.014) -0.0011 4 3 401.4001 -0.0004 5 4 401.7767 0.0031 6 5 (402.121) -0.0135 7 6 402.4864 0.0035 9 8 403.1439 0.0022 10 9 403.4454 -0.0065 11 10 (403.765) 0.0163 13 12 404.3008 -0.0027 14 13 404.5608 0.0005 15 14 404.8038 0.0005 16 15 405.2443 -0.0026 19 18 405.6313 -0.0026 19 18 405.6313 -0.0047 22 21 406.0971 0.0011 21 20 405.9524 -0.0047 22 406.2140 -0.0051 24 23 406.3273 0.0015	23	23	396.1630	0.0012
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	24	24	395.8385	0.0045
26 26 $395 \cdot 1329$ $0 \cdot 0014$ 27 27 $394 \cdot 7519$ $-0 \cdot 0041$ 28 28 $(394 \cdot 366)$ $0 \cdot 0021$ 29 29 $(393 \cdot 938)$ $-0 \cdot 0170$ 30 30 $(393 \cdot 529)$ $0 \cdot 0000$ 31 31 $(393 \cdot 071)$ $-0 \cdot 0145$ 3 2 $(401 \cdot 014)$ $-0 \cdot 0011$ 4 3 $401 \cdot 4001$ $-0 \cdot 0004$ 5 4 $401 \cdot 7767$ $0 \cdot 0031$ 6 5 $(402 \cdot 121)$ $-0 \cdot 0135$ 7 6 $402 \cdot 4864$ $0 \cdot 0035$ 9 8 $403 \cdot 1439$ $0 \cdot 0022$ 10 9 $403 \cdot 4454$ $-0 \cdot 0065$ 11 10 $(403 \cdot 765)$ $0 \cdot 0163$ 13 12 $404 \cdot 3008$ $-0 \cdot 0027$ 14 13 $404 \cdot 5608$ $0 \cdot 0005$ 15 14 $404 \cdot 8038$ $0 \cdot 0005$ 16 15 $405 \cdot 2443$ $-0 \cdot 0026$ 19 18 $405 \cdot 6313$ $-0 \cdot 0026$ 19 18 $405 \cdot 6313$ $-0 \cdot 0047$ 22 21 $406 \cdot 0971$ $0 \cdot 0011$ 21 20 $405 \cdot 9524$ $-0 \cdot 0051$ 24 23 $406 \cdot 3273$ $0 \cdot 0015$	25	25	(395-473)	-0·0175
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	26	26	395.1329	0.0014
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	27	27	394.7519	-0.0041
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	28	28	(394.366)	0.0021
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	29	29	(393.938)	-0.0170
31 31 $(393 \cdot 0/1)$ -0.0145 3 2 $(401 \cdot 014)$ -0.0011 4 3 $401 \cdot 4001$ -0.0004 5 4 $401 \cdot 7767$ 0.0031 6 5 $(402 \cdot 121)$ -0.0135 7 6 $402 \cdot 4864$ 0.0035 9 8 $403 \cdot 1439$ 0.0022 10 9 $403 \cdot 4454$ -0.0065 11 10 $(403 \cdot 765)$ 0.0163 13 12 $404 \cdot 3008$ -0.0027 14 13 $404 \cdot 5608$ 0.0005 15 14 $404 \cdot 8038$ 0.0005 16 15 $405 \cdot 0285$ -0.0038 17 16 $405 \cdot 2443$ -0.0026 19 18 $405 \cdot 6313$ -0.0009 20 19 $405 \cdot 8034$ 0.0011 21 20 $405 \cdot 9524$ -0.0047 22 21 $406 \cdot 0971$ 0.0015 24 23 $406 \cdot 3273$ 0.0015	30	30	(393.529)	0.0000
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	31	31	(393·071)	-0.0145
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3	2	(401.014)	-0.0011
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	4	3	401-4001	-0.0004
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	ے د	4	401.767	0.0031
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	07	3	(402.121)	-0.0135
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	á	0	402-4604	0.0033
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	10	. 0	403.1439	0.0022
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11	10	403-4434	-0.0003
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	13	10	(403.703)	0.0027
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	14	12	404,5608	0.0005
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	15	14	404 3008	0.0005
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	16	15	405.0285	-0.0038
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	17	16	405.2443	-0.0026
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	19	18	405-6313	-0.0009
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	20	19	405.8034	0.0011
22 21 406·0971 0·0010 23 22 406·2140 -0·0051 24 23 406·3273 0·0015 25 24 406·3157 0.00051	21	20	405.9524	-0.0047
23 22 406·2140 -0·0051 24 23 406·3273 0·0015 25 24 406·3273 0·0015	22	21	406.0971	0.0010
24 23 406·3273 0·0015 25 24 406·4157 0·0015	23	22	406.2140	-0.0051
25 24 406,4157 0,0002	24	23	406.3273	0.0015
25 24 400·4157 -0·0002	25	24	406-4157	-0.0002

Table 6 (continued).

	$(v=0, K=0)^{\dagger}$		(v=0, I)	$K = 1) \dagger \P$	$(v_6 = 1, K = 1)$ ‡	
	l(A ⁺)§	u(B ⁺)§	l(A ⁺)§	u(B ⁺)§	<i>l</i> (<i>A</i> [−])§	u(B ⁻)§
$\frac{\bar{B}/\text{cm}^{-1}}{\bar{D}/10^{-6} \text{ cm}^{-1}} \\ \frac{\bar{D}}{H/10^{-12} \text{ cm}^{-1}} \\ \frac{\bar{B}-C}{(B-C)/10^{-3} \text{ cm}^{-1}} \\ \frac{l/10^{-9} \text{ cm}^{-1}}{\tilde{v}_0/\text{cm}^{-1}} $	0·2167083 2·0538 -44·33 0 0 0	0·2165735 2·0380 -47·14 0 0 0·65869	0.217890 2.043 34.8 3.1697 77.3 35.86236** 35.85919	0.217693 1.953 34.8 3.044 77.7 36.92275 ** 36.92580	$\begin{array}{c} 0.2105_{62} \\ 2.5_{16} \\ 0 \text{ (fix)} \\ 0.7_{61} \\ -70 \\ 399.78_{65} \end{array}$	$\begin{array}{c} 0.2103_{05} \\ 2.5_{41} \\ 0 \text{ (fix)} \\ 0.5_{37} \\ -171 \\ 400.75_{34} \end{array} $

Table 7. Spectroscopic constants used in the fits.

† From [8], which uses also microwave data [3]. The constants were fixed in the fit.

[‡] This work, uncertain digits are set as indices.

§ *l* is for the lower tunnelling level, *u* for the upper, the species Γ^{vb} of the vibrational state being given in parentheses.

|| Set equal to zero in the fits for consistency of the model.

¶ These values were used for the calculation of ground state combination differences in table 5.

* Transition wavenumber from upper tunnelling level (measured band centre). The term value is 401.4121.

** Asymmetry doublet of the nearly symmetric rotor state.

The vibronic ground state of $(HF)_2$ is of A^+ species (Γ^{vb}) , whereas the upper tunnelling component has B^+ species for the trans tunnelling path (see [29]). This would assign the 399.78 band to a transition from the tunnelling ground state A^+ and 400.75 to a transition from the upper tunnelling level. This is also consistent with the better root mean square deviations for this assignment shown in table 5. The significance of the different rms values for the two assignments is borderline,



Figure 5. Detail of the experimental (upper trace) and simulated (lower trace) spectrum of $(HF)_2$ (see also figure 2 and text).

because the rotational constants of the two tunnelling components do not differ much at the level of our resolution. Higher resolution data would improve the significance, but are hardly necessary because of the symmetry assignment.

Excluding a perturbation (e.g. Coriolis coupling) which would invert the normal order of the $K_a K_c$ -levels, all of the above results indicate a type C band polarization. Only the out of plane torsional transition is of type C. This assignment is to be compared further to the alternative, namely the $v_3(A')$ high frequency fundamental. This would result in a type B rotational band polarization (possibly a hybrid). There is also a second argument against this assignment. Assuming the trans tunnelling path, following Hougen and Ohashi [29] the v_3 fundamental, which has A^+ vibrational symmetry Γ^{v} in the upper vibrational level (except tunnelling), would result in a transition from the upper tunnelling level in the vibrational ground state to the lower tunnelling level in the excited state and vice versa, similar to the symmetric HF stretching fundamental (see the left hand part of figure 7 of [29]). Excluding an inversion of the tunnelling levels in the excited vibrational state, which might, in principle, be caused by Fermi resonances and similar perturbations, the A^+ level should be lower in energy. Thus the $A^+('') \rightarrow B^+(')$ vibronic transitions should occur at high frequency compared to $B^+(") \rightarrow A^+(')$, if v_3 were the correct assignment. The opposite behaviour is observed. The v_3 assignment would thus appear to be excluded. The same kind of argument would also rule out any first overtone transition, which has $\Gamma^{v'} = A^+$, in particular the dimer valence stretching overtone.

The observed ordering of the two subband origins has a natural explanation if the assignment to the torsional fundamental is accepted. The difference of the two subband origins is here just equal to the increase of the tunnelling splitting, when going from the ground state with K'' = 0, $v_b = 0$ to the excited state with K' = 1, $v_b'' = 1$.

Table 8 summarizes the tunnelling splittings for different fundamentals from the work of Pine *et al.* [9] and the present investigation. It is seen that both rotational excitation by $\Delta K = 1$ and vibrational excitation of the torsion lead to an increase of the tunnelling splitting by about a factor of 1.5. In contrast to this, HF vibrational stretching excitation gives a *decrease* of the tunnelling splitting. Both effects can be understood with simple models. In particular, the increase of tunnelling rate with torsional excitation indicates some contribution of the out of plane motion to the interconversion of conformers. This contribution is expected but now found to be still relatively small, which would seem to justify the decision of Hougen and Ohashi to concentrate on planar tunnelling motion [29]. The relatively small tunnelling splitting of the excited state is also evidence against all assignments involving some excitation of the large amplitude bending motion, assuming that this can be associated with both the tunnelling coordinate and the predicted low frequency bending

		Δ	v/GHz	
<i>K</i> 0	v = 0 19.747	$v_1 = 1$ 6.461	$v_2 = 1$ 6.998	$v_6 = 1$
1 Reference	31·911 [9, 3]	10·488 [9]	10·225 [9]	48·7 ₇ This work

Table 8. (HF)₂ tunnelling splittings Δv for three fundamentals.

fundamental. Together with the symmetry assignment, this would rule out all possible assignments of the observed band to an overtone or combination tone.

The decrease of B_{vk} with torsional excitation is expected because of the weaker and thus longer hydrogen bond in excited torsional states. Related decreases have been found in HCN...HF [35] and rare gas ...HF complexes [36]. In (HF)₂ (B'' - B') is a factor of three larger than for (HCN...HF). Another interesting observation is the decrease of the asymmetry splitting (by a factor of 6) in the $K_a = 1$ state upon torsional excitation. The value of (B - C) is comparable to that of the quasilinear molecular HNCS [38]. A question arises concerning the intensity of the $K = 2 \leftarrow 1$ subband, which in an ordinary molecule should be about as intense as $K = 1 \leftarrow 0$. It is obvious from figure 1 that to the high wavenumber side there are only rather weak bands. This might be explained by a decrease of intensity of $K = 2 \leftarrow 1$ or by a spreading of the intensity over a wider range. Interestingly rather intense subbands occur to the *low* wavenumber side of the 400 cm⁻¹ system.

3.4. Further subbands and possible assignments

Two moderately strong subbands can be assigned near 380 cm^{-1} . The analysis of these bands is less straightforward than for the 400 cm^{-1} region. Firstly the signal to noise ratio is somewhat poorer to low wavenumbers for the Cu/Ge detector used here. This causes difficulties for the analysis of the high Q-branch transitions and for the P-branch. Secondly, the R-branch overlaps considerably with the 400 cm^{-1} bands, resulting in a very congested spectrum. The difficulty is amplified by apparent perturbations for J > 30 in the 400 cm^{-1} band. The Q-branches are, however, rather clear and figure 6 shows a detail of the structure. A list of assigned transitions is presented in table 9. There is no evidence for asymmetry splitting. Spin statistical intensity alternations are visible, but not very compelling. Due to the lack of adequate ground state combination differences, the ground state of the transition cannot be established beyond doubt. Three possible assignments are: (i) The $K = 0 \leftarrow 1$ transition of the torsion; (ii) the $K = 0 \leftarrow 1$ transition of the in plane bending; (iii) the $K = 1 \leftarrow 0$ of the in plane bending vibration. Other assignments,



Figure 6. Detail of the subband near 380 cm^{-1} (see text, resolution = 0.02 cm^{-1} FWHM, p = 2 kPa, T = 250 K, 1 = 10 m).

	Wavenu	mber/cm ⁻¹
J' = J''	ν̃exp	$(\tilde{v}_{exp} - \tilde{v}_{calc})$
8	383-1399	0.0004
9	383·0881	0.0010
10	383.0238	-0.0046
11	382.9613	-0.0050
12	382.8901	-0.0014
13	382.8132	0.0004
14	382.7274	0.0003
15	382.6385	0.0044
16	382.5321	-0.0014
17	382-4323	0.0071
18	382.3094	0.0006
19	382-1871	0.0030
20	382-0492	-0.0016
21	381.9037	-0.0048
22	381.7610	0.0041
23	381.5912	-0.0044
24	381-4244	0.0000
25	381·2392	-0.0035
27	380-8483	0.0018
29	380.4003	-0.0034
30	380.1676	0.0039

Table 9. Assigned Q-branch transitions of the subbands near 380 cm^{-1} . (a) Upper tunnelling component ($v_{sub}^0 = 383.34 \text{ cm}^{-1}$).

(b)	Lower	tunnelling	component	$(\tilde{v}_{mb}^{0} =$	= 382.028 cm	- I).

	Wavenumber/cm ⁻¹		
J' = J''	v _{exp}	$(\tilde{v}_{exp} - \tilde{v}_{calc})$	
7	381.7610	-0.0001	
8	381.6938	0.0093	
9	381.5912	-0.001	
10	381.4973	-0.0021	
11	381.3940	-0.0029	
12	381.2856	0.0040	
13	(381-142)	-0.0145	
14	(381.002)	-0·0196	
15	380.8795	0.0026	
16	(380.733)	0.0107	
17	380-5595	0.0018	
19	380.1968	-0.0018	
20	380.0064	0.0025	
21	379.7912	-0.0079	
22	379.5848	0.0008	
23	379-3634	0.0047	
24	(379.110)	-0.0130	
26	378.6193	-0.0009	

such as vibrational hot band transitions are less likely. Each of the above assignments would have interesting implications. The first assignment would assume that the energy of the K = 0, v = 1 state is higher than of K = 1, v = 1. This could result from Coriolis interaction with higher states as has been observed for HNCS [38]. One would also have to assume an inversion of the tunnelling doublet. The second assignment would imply a change of the expected intensity distributions for the $K = 0 \leftarrow 1$ and $K = 1 \leftarrow 0$ transitions, for example due to a B/C Coriolis coupling [39]. This assignment would also imply a very small tunnelling splitting of 0.26 cm^{-1} in the excited state. The third assignment is very attractive because it is consistent with the chemical intuition that the torsion and the in plane bending vibration are really a pair of nearly degenerate vibrations resulting from the zero order picture of a linear $F-H \cdots F$ molecule with a degenerate bending vibration. However, all *ab initio* calculations predict the in plane bending vibration to occur at *higher* frequency than the torsion. Thus the vibrational assignment must remain open, at present.

The rotational assignment is also still somewhat uncertain. A shift of the J assignment by (± 1) unit gives fits of practically identical quality as the assignment given in table 9. Larger shifts give systematically poorer root mean square deviations. The constants arising from the more probable assignments are summarized in table 10. The change of rotational constants for this state is about a factor of 2 smaller than for the 400 cm⁻¹ level.

4. Conclusions and outlook

It has been possible to detect and assign parts of the weak and highly complex rovibrational spectrum of the hydrogen fluoride dimer in the far infrared. A subsystem near 400 cm^{-1} can be identified with a high degree of confidence as being due to the tunnelling levels of the $K = 1 \leftarrow 0$ transition of the torsional (A") fundamental, which would then normally fall around 320 cm^{-1} . We may summarize the main arguments in favour of this assignment:

Table 10. Constants obtained from the fits of the 380 cm⁻¹ bands ($\Delta X = X' - X''$).

	(A)		
	а	b	с
$ \tilde{v}_{sub}/cm^{-1} \Delta B_{eff}/cm^{-1} \dagger \Delta D_{eff}/10^{-6} cm^{-1} \ddagger rms/cm^{-1} $	$\begin{array}{cccccccccccccccccccccccccccccccccccc$		$ \frac{383 \cdot 31_{00}}{-0.0030_{77}} \\ 0.6_{1} \\ 0.0031 $
	a	Ь	c
\tilde{v}_{sub}/cm^{-1} $\Delta B_{eff}/cm^{-1}$ $\Delta D_{eff}/10^{-6} cm^{-1}$ rms/cm^{-1}	$382.02_{85} \\ - 0.0047_{65} \\ 0.04_{6} \\ 0.0046$	$ \frac{382 \cdot 08_{07}}{-0.0043_{49}} \\ 0.2_{2} \\ 0.0049 $	$ \begin{array}{r} 381 \cdot 97_{50} \\ -0.0052_{36} \\ -0.2_{1} \\ 0.0056 \end{array} $

 $\begin{array}{l} \dagger \ B_{\rm eff} = \bar{B}_{\rm vk} \pm \delta_{\rm kl} (B-C)/4. \\ \ddagger \ D_{\rm eff} = D_{\rm vk} \pm \delta_{\rm kl} \, l_{\rm vk}/2. \end{array}$

- (i) From *ab initio* calculations the band is expected to belong to one of the two higher frequency fundamentals of the low frequency modes.
- (ii) The symmetry assignment of the tunnelling components by means of nuclear spin statistics defines the vibrational symmetry of upper and lower tunnelling levels.
- (iii) The ground state of the transition is shown to be the vibrational ground state by combination differences of assigned rotational lines and known data from I.R. and microwave spectroscopy.
- (iv) Overtone and combination transitions of the low frequency bending are excluded because of the modest increase in the tunnelling splitting. The overtone of the hydrogen bond (F-F stretching) vibration is excluded because of its symmetry.
- (v) One observes no K-type asymmetry splitting, which is consistent with a $K = 1 \leftarrow 0$ transition.

Although one might raise objections against any single argument, for instance the symmetry assignments depend somewhat upon dynamical assumptions made in the group theoretical analysis of the tunnelling process [29], the ensemble of reasons renders the assignment rather compelling. The assignment of a second observed subband near 380 cm^{-1} is still uncertain. More complete interpretation will be possible when further low frequency modes are observed and analysed. Such work is in progress in our laboratory.

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